Jing Yin, Yihui Bai, Mengyi Mao, and Gangguo Zhu*

Department of Chemistry, Zhejiang Normal University, 688 Ying[bin](#page-5-0) Road, Jinhua 321004, China

S Supporting Information

[AB](#page-5-0)STRACT: [A silver-cataly](#page-5-0)zed trans addition of carboxylic acids to ynol ethers is described. The reaction has a broad scope with respect to carboxylic acids and ynol ethers, delivering (Z) - α -alkoxy enol esters in good yields with excellent regio- and stereoselectivity. Meaningfully, the Ni-

catalyzed selective coupling of alkenyl C−OPiv bonds of (Z)-α-alkoxy enol esters with boronic acids enables a convenient route to the access of (E)-enol ethers. As such, the two-step procedure, consisted of a hydrocarboxylation and a subsequent Suzuki− Miyaura coupling, offers a formal trans hydroarylation of ynol ethers, thus providing a good complementary method to our previous report.

■ INTRODUCTION

Enol esters are versatile building blocks in organic synthesis and material science. Hence, the development of efficient and general methods for the preparation of these moieties is still of high interest. Specifically, the transition-metal-catalyzed addition of carboxylic acids to acetylenes appears to be an ideal protocol due to its high atom economy.^{1,2} Various catalysts, including mercury,³ ruthenium,⁴ rhodium,⁵ iridium,⁶ palladium,⁷ sil[ve](#page-5-0)r,⁸ gold,⁹ and rhenium,^{10,11} have been described for promoting the tr[an](#page-5-0)sformation during t[he](#page-5-0) past [de](#page-5-0)cades. How[ev](#page-5-0)er, [mo](#page-5-0)st o[f](#page-5-0) the previou[s re](#page-5-0)ports are restricted to terminal alkynes. The examples on hydrocarboxylation of unsymmetric internal acetylenes are particularly rare, $7,12-14$ mainly because of the difficulty in controlling the regio- and stereoselectivity. By means of a $Ru_3(CO)_{12}/3PPh_3$ [catal](#page-5-0)[yst](#page-6-0) system, a highly selective cis addition of carboxylic acids to aryl and trifluoromethyl group substituted unsymmetric internal alkynes has been accomplished by the groups of Kawatsura and Itoh.¹² Recently, Lam and co-workers demonstrated that carboxylic acids could add to ynamides, providing α acyl[oxy](#page-5-0)enamides in a highly regio- and stereocontrolled manner.¹³ We have initiated a project aiming at the exploration of new applications of ynol ethers,15−¹⁸ including chloroallylation,¹⁵ [h](#page-5-0)ydroarylation,¹⁶ and hydroboration¹⁷ processes. For example, a cis hydroarylation of yn[ol ethe](#page-6-0)rs with boronic acids as ary[lat](#page-6-0)ing reagents ha[s b](#page-6-0)een developed in [ou](#page-6-0)r group, giving (Z) -enol ethers in high yields (Scheme 1).^{16a} In contrast, the trans hydroarylation of ynol ethers has not been realized. Herein, we describe a silver-catalyzed [ster](#page-6-0)eospecific trans addition of carboxylic acids to ynol ethers, furnishing (Z) - α alkoxy enol esters in high yields with excellent regio- and stereoselectivity. Further, the subsequent Ni-catalyzed Suzuki− Miyaura coupling¹⁹ of alkenyl C−OPiv bonds²⁰ of (Z) - α -alkoxy enol esters enables the generation of (E) -enol ethers with excellent geome[tric](#page-6-0) purity. Notably, the t[wo-](#page-6-0)step procedure displayed here offers a formal trans hydroarylation of ynol

Scheme 1. Summary of This Work

ethers, thus providing a good complementary method to our previous report.¹⁶

■ RESULTS [AN](#page-6-0)D DISCCUSION

As a model system, the reaction between ynol ether 1a and benzoic acid 2a was first investigated. In the presence of 5 mol % of $Ru_3(CO)_{12}$, 3aa was obtained in 31% yield after stirring in toluene at 100 °C for 10 h, and no other regio- and stereoisomers were observed by GC and NMR analysis (Table 1, entry 1). Further optimization of the catalyst indicated that $Ag₂O$ served as the preferred catalyst for this reaction, [a](#page-1-0)ffording 3aa in 65% isolated yield (Table 1, entries 1−12). Pleasingly, the utilization of dioxane instead of toluene as the solvent increased the yield to 88%, whereas th[e](#page-1-0) reaction performed under an atmosphere of air resulted in a relatively lower yield (Table 1, entries 17 and 18). Therefore, the optimized conditions for regio- and stereoselective addition of carboxylic acids to y[no](#page-1-0)l ethers comprised 5 mol % of Ag2O as the catalyst, dioxane as the solvent, and at 100 $^{\circ}$ C under N₂ atmosphere for 10 h.

Under the optimized reaction conditions, we observed a broad scope with respect to benzoic acids (Table 2). In particular, 4-fluoro, 4-chloro, and 4-bromobenzoic acids

Received: July 17, 2014 Published: September 4, 2014

Table 1. Optimization of the Reaction Conditions a

n -Bu \rightleftharpoons	-O-3-tol + PhCOOH 1a 2a	[M] solvent, 100 °C	n-Bu OBz O-3-tol н Заа
entry	[M]	solvent	yield $(\%)^b$
1	$Ru_3(CO)_{12}$	toluene	31
2	ZnCl ₂	toluene	trace
3	CuCl	toluene	12
$\overline{4}$	Cu(OAc)	toluene	17
5	Na ₂ CO ₃	toluene	trace
6	Ag_2O	toluene	65
7	Ag_2CO_3	toluene	53
8	AgNO ₃	toluene	20
9	AgOAc	toluene	18
10	AgOTf	toluene	trace
11	PPh ₃ AuCl/AgOTf	toluene	trace
12	$PPh_3AuCl/AgSbF_6$	toluene	trace
13	Ag_2O	THF	69
14	Ag_2O	ClCH ₂ CH ₂ Cl	82
15	Ag ₂ O	DMF	51
16	Ag_2O	NMP	63
17	Ag_2O	dioxane	88
18 ^c	Ag_2O	dioxane	77
$\tau-$	٠.	\sim \sim \sim \sim	

 a Reaction conditions: 1a (0.25 mmol), 2a (0.28 mmol), [M] (5 mol %), solvent (1 mL), under N_2 , 100 °C, 10 h. b^b Isolated yield. CUnder air atmosphere. Bz = benzoyl.

participated well in the present reaction, leaving the halide atoms untouched (Table 2, entries 1−3). However, reaction with 4-iodobenzoic acid as the substrate was sluggish under the same conditions; fortunately, the utilization of 1.2 equiv of $Et₃N$ as the additive significantly improved the yield (Table 2, entry 4). A wide range of functional groups, including $NO₂$, CN, $CO₂Me$, Me, OMe, NH₂, OH, pyridyl, and thienyl functionalities, survived well under the reaction conditions, producing (Z) - α -alkoxy enol esters 3 in moderate to excellent yields with perfect regio- and stereocontrol, although the addition of Et_3N was also necessary for the transformation of $2j$ and 2n (Table 2, entries 5−14).

Besides benzoic acids, other carboxylic acids were also evaluated. For example, cinnamic acid gave rise to (Z) - α -alkoxy enol ester 3ap in 85% yield under the reaction conditions (Table 2, entry 15). Phenylpropiolic acid $(2q)$ was found to be almost unreactive under the standard conditions, probably due to the competitive interaction of the C−C triple bond of 2q with the silver catalyst. We envisioned that increasing the amounts of $Ag₂O$ might be able to facilitate the reaction. As expected, the use of 2.5 equiv of Ag_2O together with the addition of $Et₃N$ (1.2 equiv) led to the desired product in 73% yield (Table 2, entry 16). As for HOAc, an excess use (5 equiv) of starting material was required for the full conversion (Table 2, entry 17). We were pleased to find that the reaction of pivalic acid with 1a proceeded smoothly as well, providing the corresponding product in satisfactory yield (Table 2, entry 18). It implied that the steric hindrance of carboxylic acid has little impact on the reaction yield. Furthermore, the silver-catalyzed trans addition of 4-fluorophenylacetic acid to 1a occurred uneventfully, delivering 3at in 76% yield (Table 2, entry 19). In contrast, no desired products were obtained when CF_3SO_3H or $CF₃CO₂H$ was used, owing to the hydrolysis of ynol ethers in these cases.

Table 2. Scope of Silver-Catalyzed Trans Addition of Carboxylic Acids to Ynol Ethers^a

^aReaction conditions: 1 (0.25 mmol), 2 (0.28 mmol), Ag₂O (5 mol %), dioxane (1 mL), N_2 , 100 °C, 10 h. b Isolated yield. c Et₃N (0.3) mmol) was added to the standard conditions. d_{1a} (0.25 mmol), 2n (0.5 mmol), Ag₂O (5 mol %), Et₃N (0.6 mmol), dioxane (1 mL), N₂, 100 °C, 24 h. e Ag₂O (0.7 mmol) and Et₃N (0.3 mmol) were used. From $\frac{f_{\text{Run}}}{f}$ at room temperature for 4 h.

On the other hand, various ynol ethers were successfully applied to the current reaction. For instance, terminal ynol ether 1b furnished 3ba in 90% yield (Table 2, entry 20). The addition of 2a to 1d, a substrate bearing both C−C triple and C−C double bonds, underwent smoothly to form 3da in 87% yield with excellent chem-, regio-, and stereoselectivity (Table 2, entry 22). Moreover, increased steric bulk of the $R¹$ group was well-tolerated, giving 3fa in 82% yield with excellent regioand stereocontrol (Table 2, entry 26). Other than alkyl ynol ethers, aromatic substrate lk produced hydrocarboxylation products 3ka and 3ks in respective yields of 95% and 87% (Table 2, entries 31 and 32). Notably, the electronic effect of the $R¹$ group has no significant impact on the reaction

efficiency, as demonstrated by the production of 3la and 3ma (Table 2, entries 33 and 34). Remarkably, the commercially available ethyl ethynyl ether was also an effective substrate, giving [3n](#page-1-0)a in 71% yield after reaction at room temperature in the presence of 1.2 equiv of Et_3N for 4 h (Table 2, entry 35). The regio- and stereochemistry of this silver-catalyzed trans hydrocarboxylation reaction was determined b[y](#page-1-0) the NOE measurements (see the Supporting Information).

A plausible mechanism for this reaction is proposed as shown in Scheme 2. First, the e[lectrophilic addition of a](#page-5-0) silver cation to

Scheme 2. A Possible Mechanism

C−C triple bonds of ynol ethers forms an intermediate I, which can be attacked by the carboxylate anion from the backside to give a silver species II. The polarization of C−C triple bonds of ynol ethers stemmed from the electron donation of the oxygen atom may account for the regioselective α -addition of the carboxylate anion.15−¹⁸ Finally, protodemetalation of the C−Ag bond of II delivers the trans addition products 3 along with the regeneration of t[he](#page-6-0) s[ilv](#page-6-0)er catalyst (Scheme 2).

As such, we have developed a direct, efficient, and ligand-free method for the production of (Z) - α -alkoxy enol esters in good yields with perfect regio- and stereoselectivity. We noticed that products 3 have two different alkenyl C−O bonds. We envisioned that the selective cross-coupling of C−OCOR³ bonds over C−OR² bonds might result in a stereocontrolled approach to enol ethers. To this end, pivalate 3ks was treated with 4 mol % of $Ni(PCy_3)_2Cl_2$, 1.2 equiv of $PhB(OH)_2$ (4a), and 3.0 equiv of K_2CO_3 in toluene at 130 °C for 12 h; as expected, (E) -enol ether $5a^{21}$ was obtained in 35% yield. Other bases such as KOt-Bu, NaOMe, and KF generated 5a in 10, 66, and 5% yield, respectively. [In](#page-6-0) contrast, the utilization of K_3PO_4 as the base led to the highest yield (86%), and the coupling of the C−OEt bond was not detected (Table 3, entry 1). Then, a brief survey on the scope of boronic acids was conducted. Both 4- and 2-tolylboronic acids worked well for this Suzuki− Miyaura coupling reaction (Table 3, entries 3 and 4). Moreover, 4e, an electron-poor boronic acid, underwent the

Table 3. Synthesis of (E)-Enol Ethers via the Ni-Catalyzed C $-$ O Bond Coupling^a

R^1 R^1 OPiv Ar $Ni(PCy_3)_2Cl_2$, ArB(OH) ₂ (4), K ₃ PO ₄				
	OR ² н	toluene, 130 °C, 12 h н	OR ² 5	
entry	R^{1}/R^{2} (3)	Ar (4)	yield $(\%)^b$	
$\mathbf{1}$	Ph/Et(3ks)	Ph $(4a)$	86(5a)	
$\overline{2}$	Ph/Et(3ks)	4-FC ₆ H ₄ (4 b)	71(5b)	
3	Ph/Et(3ks)	4-Me C_6H_4 (4c)	74 $(5c)$	
4	Ph/Et(3ks)	2-Me C_6H_4 (4d)	70(5d)	
5	Ph/Et(3ks)	$4-CF_3C_6H_4$ (4e)	83 (5e)	
6	$n-Bu/3$ -tol $(3as)$	Ph $(4a)$	60(5f)	
7	allyl $/3$ -tol $(3ds)$	Ph $(4a)$	63(5g)	
8	$H/3$ -tol $(3es)$	Ph $(4a)$	50(5h)	

^aReaction conditions: 3 (0.25 mmol), 4 (0.3 mmol), $Ni(PCy₃)₂Cl₂$ (4 mol %), K₃PO₄ (0.75 mmol), toluene (1 mL), under N₂, 130 °C, 12 h. ^bIsolated yield.

facile coupling with 3ks to form (E) -enol ether 5e in 83% yield (Table 3, entry 5). In addition to 3ks, other pivalates, including 3as, 3ds, and 3es, were found to be the suitable coupling partners and led to the desired products in reasonable yields (Table 3, entries 6−8). Overall, the procedure involving hydrocarboxylation and subsequent Suzuki−Miyaura coupling offers a formal trans hydroarylation of ynol ethers, which provides a good complementary method to our previous efforts to enol ethers.¹⁶

■ CONCLU[SI](#page-6-0)ON

In summary, a silver-catalyzed trans addition of carboxylic acids to ynol ethers has been realized, which produces (Z) -α-alkoxy enol esters in good yields with excellent regio- and stereoselectivity. A wide range of functional substituents, such as NO₂, CN, CO₂Me, Me, OMe, NH₂, OH, pyridyl, thienyl, alkenyl, and alkynyl groups, are well-tolerated in this reaction. Furthermore, the subsequent Ni-catalyzed Suzuki−Miyaura coupling of alkenyl C−OPiv bonds of (Z) - α -alkoxy enol esters enables a convenient approach to $\mathcal (E)$ -enol ethers. As such, the two-step procedure²² developed here offers a formal *trans* hydroarylation of ynol ethers, thereby providing a good complementary me[tho](#page-6-0)d to our previous report. We believe that it will be valuable for the diversity-oriented synthesis.²³

EXPERIMENTAL SECTION

General. Unless otherwise mentioned, materials obtained from commercial suppliers were used directly without further purification. Toluene, dioxane, and THF were distilled from sodium prior to use. Column chromatography was performed using silica gel (300−400 mesh) with petroleum ether/EtOAc as the eluent. ¹H NMR and ¹³C NMR spectra were measured on a 600 MHz NMR spectrometer using CDCl₃ or DMSO-d6 as the solvent. Chemical shifts were given in δ relative to TMS, and the coupling constants were given in Hz. Highresolution mass spectra (HRMS) analyses were conducted using a TOF MS instrument with an ESI or APCI source.

General Procedure for the Ag-Catalyzed Regio- and Stereoselective Addition of Carboxylic Acids to Ynol Ethers. To a mixture of $2a$ (34 mg, 0.28 mmol) and Ag₂O (3 mg, 0.0125 mmol) was added a solution of 1a (47 mg, 0.25 mmol) in 1 mL of dioxane under a nitrogen atmosphere. After stirring at 100 °C for 10 h, the reaction mixture was quenched with water, extracted with EtOAc, washed with brine, dried over anhydrous $Na₂SO₄$, and concentrated. Column chromatography on silica gel (petroleum ether/EtOAc = 30:1) gave 68 mg (yield: 88%) of 3aa as a colorless oil. $^1{\rm H}$ NMR (600 MHz, CDCl₃): δ 8.10–8.05 (m, 2H), 7.58 (t, J = 7.4 Hz, 1H), 7.46– 7.43 (m, 2H), 7.22−7.18 (m, 1H), 7.03−6.98 (m, 2H), 6.89 (d, J = 7.4 Hz, 1H), 4.81 (t, $J = 7.6$ Hz, 1H), 2.33 (s, 3H), 2.05–2.01 (m, 2H), 1.40−1.30 (m, 4H), 0.88 (t, J = 7.2 Hz, 3H); ¹³C NMR (151 MHz, CDCl3): δ 163.6, 155.6, 148.5, 139.7, 133.7, 130.2, 129.2, 128.5, 128.5, 124.3, 118.5, 114.6, 100.4, 31.4, 24.5, 22.1, 21.3, 13.8; HRMS (ESI) calcd for $C_{20}H_{22}NaO_3$ $(M + Na)^+$ 333.1467, found 333.1461.

Compound 3ab. 62 mg, 75% yield, colorless oil; ¹H NMR (600 MHz, CDCl3): δ 8.15−8.04 (m, 2H), 7.22−7.18 (m, 1H), 7.15−7.10 $(m, 2H)$, 7.00−6.98 $(m, 2H)$, 6.90 $(d, J = 7.5 Hz, 1H)$, 4.80 $(t, J = 7.6$ Hz, 1H), 2.34 (s, 3H), 2.04−1.99 (m, 2H), 1.41−1.30 (m, 4H), 0.88 (t, $J = 7.2$ Hz, 3H); ¹³C NMR (151 MHz, CDCl₃): δ 166.2 (d, $J =$ 255.6 Hz), 162.6, 155.5, 148.4, 139.8, 132.9 (d, J = 9.5 Hz), 129.2, 124.8 (d, $J = 2.8$ Hz), 124.4, 118.6, 115.8 (d, $J = 22.1$ Hz), 114.7, 100.4, 31.4, 24.5, 22.1, 21.4, 13.8; HRMS (ESI) calcd for $C_{20}H_{21}FNaO₃$ $(M + Na)⁺ 351.1372$, found 351.1371.

Compound 3ac. 70 mg, 81% yield, colorless oil; ¹H NMR (600 MHz, CDCl3): δ 7.97−7.90 (m, 2H), 7.38−7.31 (m, 2H), 7.13−7.10 $(m, 1H)$, 6.94–6.80 $(m, 2H)$, 6.82 $(d, J = 7.5 Hz, 1H)$, 4.72 $(t, J = 7.6$ Hz, 1H), 2.25 (s, 3H), 1.97−1.91 (m, 2H), 1.33−1.22 (m, 4H), 0.79 $(t, J = 7.2 \text{ Hz}, 3\text{H})$; ¹³C NMR (151 MHz, CDCl₃): δ 162.8, 155.5, 148.4, 140.4, 139.7, 131.6, 129.2, 128.9, 127.0, 124.5, 118.6, 114.7,

The Journal of Organic Chemistry **Article Article Article Article Article Article Article Article Article**

100.4, 31.4, 24.5, 22.1, 21.4, 13.8; HRMS (ESI) calcd for $C_{20}H_{21}$ - $CINaO₃$ $(M + Na)⁺$ 367.1077, found 367.1077.

Compound **3ad.** 75 mg, 77% yield, colorless oil; ¹H NMR (600 MHz, CDCl₃): δ 7.93 (d, J = 8.5 Hz, 2H), 7.58 (d, J = 8.5 Hz, 2H), 7.21−7.17 (m, 1H), 7.00−6.97 (m, 2H), 6.89 (d, J = 7.5 Hz, 1H), 4.81 (t, J = 7.6 Hz, 1H), 2.33 (s, 3H), 2.04−1,98 (m, 2H), 1.39−1.30 (m, 4H), 0.87 (t, J = 7.2 Hz, 3H); ¹³C NMR (151 MHz, CDCl₃): δ 162.9, 155.4, 148.4, 139.7, 131.9, 131.6, 129.2, 129.0, 127.4, 124.4, 118.5, 114.7, 100.4, 31.4, 24.5, 22.1, 21.3, 13.8; HRMS (ESI) calcd for $C_{20}H_{21}BrNaO₃$ $(M + Na)⁺$ 411.0572, found 411.0567.

Compound **3ae**. 80 mg, 73% yield, colorless oil; ¹H NMR (600 MHz, CDCl3): δ 7.87−7.84 (m, 2H), 7.83−7.80 (m, 2H), 7.26−7.22 $(m, 1H)$, 7.06–7.02 $(m, 2H)$, 6.94 $(d, J = 7.5 Hz, 1H)$, 4.85 $(t, J = 7.6$ Hz, 1H), 2.38 (s, 3H), 2.08−2.02 (m, 2H), 1.44−1.36 (m, 4H), 0.92 (t, J = 7.2 Hz, 3H); ¹³C NMR (151 MHz, CDCl₃): δ 163.1, 155.4, 148.4, 139.7, 137.9, 131.5, 129.2, 128.0, 124.4, 118.5, 114.7, 101.9, 100.4, 31.4, 24.5, 22.1, 21.3, 13.8; HRMS (ESI) calcd for $C_{20}H_{21}NaO_3$ $(M + Na)^+$ 459.0433, found 459.0426.

Compound 3af. 53 mg, 60% yield, colorless oil; 1 H NMR (600 MHz, CDCl₃): δ 8.30 (d, J = 8.8 Hz, 2H), 8.26–8.23 (m, 2H), 7.23– 7.19 (m, 1H), 7.01−6.97 (m, 2H), 6.92 (d, J = 7.5 Hz, 1H), 4.84 (t, J = 7.6 Hz, 1H), 2.34 (s, 3H), 2.07−1.98 (m, 2H), 1.41−1.31 (m, 4H), 0.88 (t, J = 7.2 Hz, 3H); ¹³C NMR (151 MHz, CDCl₃): δ 161.8, 155.3, 150.9, 148.2, 139.9, 133.9, 131.3, 129.3, 124.7, 123.7, 118.5, 114.6, 100.6, 31.4, 24.5, 22.1, 21.3, 13.8; HRMS (ESI) calcd for $C_{20}H_{21}$ -NNaO₅ (M + Na)⁺ 378.1317, found 378.1309.

Compound 3ag. 73 mg, 87% yield, colorless oil; $^1\rm H$ NMR (600 MHz, CDCl₃): δ 8.20 (d, \bar{J} = 8.0 Hz, 2H), 7.79 (d, \bar{J} = 8.0 Hz, 2H), 7.26−7.22 (m, 1H), 7.04−7.00 (m, 2H), 6.94 (d, J = 7.5 Hz, 1H), 4.86 (t, J = 7.6 Hz, 1H), 2.37 (s, 3H), 2.10−2.00 (m, 2H), 1.44−1.32 (m, 4H), 0.91 (t, J = 7.1 Hz, 3H); ¹³C NMR (151 MHz, CDCl₃): δ 162.0, 155.3, 148.2, 139.8, 132.3, 132.3, 130.6, 129.2, 124.6, 118.4, 117.6, 117.1, 114.6, 100.5, 31.3, 24.4, 22.0, 21.3, 13.7; HRMS (ESI) calcd for $C_{21}H_{21}NNaO_3$ $(M + Na)^+$ 358.1419, found 358.1414.

Compound 3ah. 77 mg, 84% yield, colorless oil; ¹H NMR (600 MHz, CDCl₃): δ 8.20–8.10 (m, 4H), 7.26–7.21 (m, 1H), 7.05–7.01 $(m, 2H)$, 6.94 (d, J = 7.5 Hz, 1H), 4.85 (t, J = 7.6 Hz, 1H), 3.98 (s, 3H), 2.37 (s, 3H), 2.10−2.00 (m, 2H), 1.45−1.30 (m, 4H), 0.91 (t, J = 7.2 Hz, 3H); ¹³C NMR (151 MHz, CDCl₃): δ 166.0, 162.8, 155.4, 148.4, 139.7, 134.6, 132.3, 130.1, 129.6, 129.2, 124.5, 118.5, 114.7, 100.4, 52.4, 31.4, 24.5, 22.1, 21.3, 13.8; HRMS (ESI) calcd for $C_{22}H_{24}NaO₅$ $(M + Na)⁺$ 391.1521, found 391.1521.

Compound 3ai. 66 mg, 81% yield, colorless oil; ¹H NMR (400 MHz, CDCl₃): δ 8.00 (d, J = 8.2 Hz, 2H), 7.30–7.21 (m, 3H), 7.06– 7.00 (m, 2H), 6.92 (d, $J = 7.5$ Hz, 1H), 4.83 (t, $J = 7.6$ Hz, 1H), 2.44 (s, 3H), 2.37 (s, 3H), 2.09−2.01 (m, 2H), 1.42−1.32 (m, 4H), 0.91 (t, $J = 7.1$ Hz, 3H); ¹³C NMR (101 MHz, CDCl₃): δ 163.6, 155.6, 148.6, 144.7, 139.7, 130.3, 129.2, 129.2, 125.8, 124.3, 118.6, 114.7, 100.3, 31.5, 24.5, 22.1, 21.7, 21.4, 13.8; HRMS (ESI) calcd for $C_{21}H_{24}NaO_3$ $(M + Na)^+$ 347.1623, found 347.1622.

Compound 3aj. 68 mg, 80% yield, colorless oil; ¹H NMR (600 MHz, CDCl₃): δ 8.09-8.04 (m, 2H), 7.25-7.20 (m, 1H), 7.05-7.00 $(m, 2H)$, 6.98–6.90 $(m, 3H)$, 4.82 $(t, J = 7.6 \text{ Hz}, 1H)$, 3.89 $(s, 3H)$, 2.37 (s, 3H), 2.08–2.03 (m, 2H), 1.43–1.35 (m, 4H), 0.91 (t, $J = 7.2$ Hz, 3H); ¹³C NMR (151 MHz, CDCl₃): δ 164.0, 163.3, 155.7, 148.6, 139.6, 132.4, 129.2, 124.3, 120.8, 118.6, 114.7, 113.8, 100.3, 55.5, 31.5, 24.5, 22.1, 21.4, 13.8; HRMS (ESI) calcd for $C_{21}H_{24}NaO_4 (M + Na)^+$ 363.1572, found 363.1565.

Compound 3ak. 33 mg, 40% yield, white solid, mp: 86–89 °C; ¹H NMR (600 MHz, CDCl₃): δ 7.91 (d, J = 8.5 Hz, 2H), 7.25–7.20 (m, 1H), 7.05−7.00 (m, 2H), 6.91 (d, J = 7.4 Hz, 1H), 6.65 (d, J = 8.5 Hz, 2H), 4.80 (t, J = 7.6 Hz, 1H), 4.19 (s, 2H), 2.36 (s, 3H), 2.07−2.01 (m, 2H), 1.42−1.33 (m, 4H), 0.90 (t, J = 7.0 Hz, 3H); 13C NMR (151 MHz, CDCl₃): δ 163.6, 155.7, 151.6, 148.7, 139.6, 132.4, 129.1, 124.1, 118.6, 117.6, 114.7, 113.7, 100.2, 31.5, 24.5, 22.1, 21.4, 13.8; HRMS (ESI) calcd for $C_{20}H_{24}NO_3$ $(M + H)^+$ 326.1756, found 326.1749.

Compound 3al. 65 mg, 80% yield, white solid, mp: 63–65 °C; ¹H NMR (600 MHz, CDCl₃): δ 8.02 (d, J = 8.8 Hz, 2H), 7.26–7.20 (m, 1H), 7.04−7.00 (m, 2H), 6.98−6.92 (m, 3H), 4.86 (t, J = 7.6 Hz, 1H), 2.36 (s, 3H), 2.10−2.00 (m, 2H), 1.50−1.30 (m, 4H), 0.91 (t, $J = 7.2$

Hz, 3H); $^{13}\mathrm{C}$ NMR (151 MHz, CDCl₃): δ 164.2, 161.4, 155.5, 148.4, 139.7, 132.7, 129.2, 124.4, 120.1, 118.5, 115.6, 114.6, 100.6, 31.4, 24.4, 22.1, 21.3, 13.8; HRMS (ESI) calcd for $C_{20}H_{22}NaO₄$ $(M + Na)⁺$ 349.1416, found 349.1414.

Compound 3am. 62 mg, 76% yield, colorless oil; ¹H NMR (600 MHz, CDCl₃): δ 10.42 (s, 1H), 7.93 (dd, J = 8.0, 1.5 Hz, 1H), 7.56– 7.51 (m, 1H), 7.28−7.23 (m, 1H), 7.07−7.02 (m, 3H), 6.96−6.93 (m, 2H), 4.89 (t, J = 7.6 Hz, 1H), 2.39 (s, 3H), 2.10−2.06 (m, 2H), 1.45− 1.37 (m, 4H), 0.93 (t, J = 7.2 Hz, 3H); ¹³C NMR (151 MHz, CDCl₃): δ 167.5, 162.2, 155.4, 147.8, 139.8, 136.6, 130.3, 129.3, 124.6, 119.4, 118.4, 117.8, 114.6, 111.0, 100.9, 31.4, 24.4, 22.1, 21.4, 13.8; HRMS (ESI) calcd for $C_{20}H_{22}NaO_4 (M + Na)^+$ 349.1416, found 349.1396.

Compound 3an. 44 mg, 56% yield, yellow oil; ¹H NMR (600 MHz, CDCl₃): δ 8.84 (d, J = 3.2 Hz, 2H), 7.91 (d, J = 5.1 Hz, 2H), 7.26– 7.22 (m, 1H), 7.02−6.99 (m, 2H), 6.94 (d, J = 7.5 Hz, 1H), 4.86 (t, J = 7.6 Hz, 1H), 2.37 (s, 3H), 2.06−2.01 (m, 2H), 1.43−1.33 (m, 4H), 0.91 (t, J = 7.2 Hz, 3H); ¹³C NMR (151 MHz, CDCl₃): δ 162.2, 155.3, 150.7, 148.2, 139.9, 135.9, 129.3, 124.7, 123.2, 118.5, 114.6, 100.7, 31.4, 24.5, 22.1, 21.4, 13.8; HRMS (ESI) calcd for $C_{19}H_{21}NNaO_3$ (M + Na)⁺ 334.1419, found 334.1417.

Compound 3ao. 70 mg, 88% yield, colorless oil; ¹H NMR (600 MHz, CDCl₃): δ 7.92 (d, \bar{J} = 3.7 Hz, 1H), 7.65 (d, J = 4.9 Hz, 1H), 7.26−7.22 (m, 1H), 7.18−7.13 (m, 1H), 7.04−7.00 (m, 2H), 6.94 (d, J $= 7.4$ Hz, 1H), 4.83 (t, J = 7.6 Hz, 1H), 2.38 (s, 3H), 2.11–2.05 (m, 2H), 1.44−1.36 (m, 4H), 0.93 (t, J = 7.1 Hz, 3H), ¹³C NMR (151) MHz, CDCl₃): δ 159.0, 155.6, 148.4, 139.8, 135.1, 133.9, 131.8, 129.3, 128.1, 124.4, 118.6, 114.8, 100.6, 31.5, 24.5, 22.2, 21.4, 13.9; HRMS (ESI) calcd for $C_{18}H_{20}NaO_3S (M + Na)^+$ 339.1031, found 339.1029.

Compound $3ap$. 71 mg, 85% yield, colorless oil; ¹H NMR (600 MHz, CDCl₃): δ 7.84 (d, J = 16.0 Hz, 1H), 7.59–7.54 (m, 2H), 7.45– 7.40 (m, 3H), 7.26−7.23 (m, 1H), 7.04−7.01 (m, 2H), 6.95 (d, J = 7.6 Hz, 1H), 6.52 (d, J = 16.0 Hz, 1H), 4.83 (t, J = 7.6 Hz, 1H), 2.39 (s, 3H), 2.10–2.04 (m, 2H), 1.44–1.38 (m, 4H), 0.95 (t, J = 7.1 Hz, 3H); ¹³C NMR (151 MHz, CDCl₃): δ 163.8, 155.7, 148.4, 147.4, 139.7, 134.0, 130.9, 129.3, 129.0, 128.4, 124.4, 118.5, 116.2, 114.8, 100.6, 31.6, 24.6, 22.3, 21.4, 13.9; HRMS (ESI) calcd for $C_{22}H_{24}NaO_3$ (M + Na)⁺ 359.1623, found 359.1626.

Compound 3aq. 61 mg, 73% yield, yellow oil; 1 H NMR (600 MHz, CDCl3): δ 8.12−8.10 (m, 2H), 7.66−7.60 (m, 1H), 7.52−7.46 (m, 2H), 7.25−7.21 (m, 1H), 7.09−7.00 (m, 3H), 4.84 (t, J = 7.6 Hz, 1H), 2.37 (s, 3H), 2.08–2.04 (m, 2H), 1.43–1.36 (m, 4H), 0.91 (t, $J = 7.2$ Hz, 3H); ¹³C NMR (151 MHz, CDCl₃): δ 163.6, 155.6, 148.6, 139.7, 133.8, 130.2, 129.2, 128.6, 128.5, 126.5, 124.4, 122.2, 118.6, 114.7, 100.4, 31.5, 24.5, 22.2, 21.4, 13.8; HRMS (ESI) calcd for $C_{22}H_{22}NaO_3$ $(M + Na)^+$ 357.1467, found 357.1463.

Compound 3ar. 52 mg, 83% yield, colorless oil; 1 H NMR (600 MHz, CDCl₃): δ 7.22 (t, J = 7.7 Hz, 1H), 6.96–6.91 (m, 3H), 4.74 (t, J = 7.6 Hz, 1H), 2.37 (s, 3H), 2.18 (s, 3H), 2.02−1.96 (m, 2H), 1.42− 1.35 (m, 4H), 0.93 (t, J = 7.1 Hz, 3H); ¹³C NMR (151 MHz, CDCl₃): δ 167.8, 155.5, 148.2, 139.7, 129.2, 124.3, 118.4, 114.6, 100.3, 31.4, 24.5, 22.2, 21.4, 20.4, 13.8; HRMS (ESI) calcd for $C_{15}H_{20}NaO_3$ (M + Na)⁺ 271.1310, found 271.1304.

Compound 3as. 58 mg, 80% yield, colorless oil; 1 H NMR (600 MHz, CDCl₃): δ 7.21 (t, J = 7.8 Hz, 1H), 6.97–6.91 (m, 3H), 4.72 (t, J = 7.6 Hz, 1H), 2.36 (s, 3H), 2.01−1.95 (m, 2H), 1.43−1.33 (m, 4H), 1.24 (s, 9H), 0.93 (t, J = 7.1 Hz, 3H); ¹³C NMR (151 MHz, CDCl₃): δ 175.3, 155.3, 148.7, 139.5, 129.1, 124.3, 118.8, 115.0, 99.3, 38.9, 31.4, 26.1, 24.3, 22.1, 21.3, 13.8; HRMS (ESI) calcd for $C_{18}H_{26}NaO_3$ (M + Na)⁺ 313.1780, found 313.1774.

Compound 3at. 65 mg, 76% yield, colorless oil; ¹H NMR (600 MHz, CDCl3): δ 7.25−7.19 (m, 3H), 7.02−6.98 (m, 2H), 6.95−6.85 $(m, 3H)$, 4.74 $(t, J = 7.7$ Hz, 1H), 3.70 $(s, 2H)$, 2.35 $(s, 3H)$, 1.90– 1.86 (m, 2H), 1.35−1.28 (m, 4H), 0.90 (t, J = 7.1 Hz, 3H); ¹³C NMR $(151 \text{ MHz}, \text{CDCl}_3): \delta 168.3, 162.1 \text{ (d, } J = 246.0 \text{ Hz}), 155.2, 148.1,$ 139.6, 130.8 (d, J = 8.1 Hz), 129.2, 128.7 (d, J = 3.3 Hz), 124.4, 118.5, 115.4 (d, J = 21.5 Hz), 114.7, 100.1, 39.9, 31.3, 24.3, 22.1, 21.3, 13.7; HRMS (ESI) calcd for $C_{21}H_{23}FNaO₃$ $(M + Na)⁺$ 365.1529, found 365.1531.

Compound 3ba. 54 mg, 90% yield, colorless oil; 1 H NMR (600 MHz, CDCl₃): δ 8.16–8.11 (m, 2H), 7.64 (t, J = 7.4 Hz, 1H), 7.51– 7.48 (m, 2H), 7.42−7.39 (m, 2H), 7.32−7.29 (m, 2H), 7.20 (t, J = 7.4 Hz, 1H), 4.42 (d, J = 3.3 Hz, 1H), 4.23 (d, J = 3.3 Hz, 1H); ¹³C NMR (151 MHz, CDCl3): δ 163.7, 156.0, 154.5, 133.8, 130.2, 129.7, 128.5, 128.5, 124.6, 119.4, 80.7; HRMS (ESI) calcd for $C_{15}H_{12}NaO_3$ (M + Na)+ 263.0684, found 263.0681.

Compound 3ca. 76 mg, 85% yield, colorless oil; ¹H NMR (600 MHz, CDCl₃): δ 8.11 (d, J = 7.7 Hz, 2H), 7.66 (t, J = 7.3 Hz, 1H), 7.53−7.50 (m, 2H), 7.36−7.33 (m, 2H), 7.28−7.22 (m, 4H), 7.02− 6.93 (m, 3H), 4.90 (t, J = 7.6 Hz, 1H), 2.80 (t, J = 7.5 Hz, 2H), 2.44– 2.38 (m, 5H); ¹³C NMR (151 MHz, CDCl₃): δ 163.5, 155.3, 148.9, 141.3, 139.6, 133.8, 130.2, 129.2, 128.5, 128.5, 128.4, 128.3, 125.9, 124.4, 118.5, 114.7, 99.2, 35.5, 26.7, 21.3; HRMS (ESI) calcd for $C_{24}H_{22}NaO_3$ $(M + Na)^+$ 381.1467, found 381.1465.

Compound 3da. 64 mg, 87% yield, colorless oil; ¹H NMR (600 MHz, CDCl₃): δ 8.14 (dd, J = 8.3, 1.1 Hz, 2H), 7.67–7.61 (m, 1H), 7.51−7.48 (m, 2H), 7.28−7.24 (m, 1H), 7.09−7.07 (m, 2H), 6.97 (d, J = 7.5 Hz, 1H), 5.90−5.80 (m, 1H), 5.14−5.10 (m, 1H), 5.05−5.03 (m, 1H), 4.85 (t, J = 7.6 Hz, 1H), 2.87–2.78 (m, 2H), 2.39 (s, 3H); ¹³C NMR (151 MHz, CDCl₃): δ 163.5, 155.2, 149.7, 139.7, 135.9, 133.8, 130.2, 129.2, 128.5, 128.4, 124.7, 119.0, 115.2, 115.2, 96.8, 29.0, 21.3; HRMS (ESI) calcd for $C_{19}H_{18}NaO_3$ (M + Na)⁺ 317.1154, found 317.1148.

Compound 3ds. 59 mg, 86% yield, colorless oil; 1 H NMR (600 MHz, CDCl₃): δ 7.23–7.20 (m, 1H), 6.97–6.92 (m, 3H), 5.85–5.78 $(m, 1H)$, 5.09 (dd, J = 17.1, 1.6 Hz, 1H), 5.02 (dd, J = 10.1, 1.4 Hz, 1H), 4.70 (t, J = 7.6 Hz, 1H), 2.73−2.70 (m, 2H), 2.36 (s, 3H), 1.23 $(s, 9H)$; ¹³C NMR (151 MHz, CDCl₃): δ 175.3, 155.0, 149.8, 139.6, 136.0, 129.2, 124.7, 119.2, 115.5, 115.1, 95.9, 39.0, 28.9, 26.8, 21.3; HRMS (APCI) calcd for $C_{17}H_{23}O_3$ $(M + H)^+$ 275.1647, found 275.1652.

Compound 3ea. 51 mg, 80% yield, colorless oil; 1 H NMR (600 MHz, CDCl₃): δ 8.16–8.10 (m, 2H), 7.64 (t, J = 7.4 Hz, 1H), 7.52– 7.49 (m, 2H), 7.28 (t, J = 7.8 Hz, 1H), 7.13–7.08 (m, 2H), 7.02 (d, J = 7.5 Hz, 1H), 4.39 (d, $J = 3.3$ Hz, 1H), 4.20 (d, $J = 3.3$ Hz, 1H), 2.40 (s, 3H); ¹³C NMR (151 MHz, CDCl₃): δ 163.8, 156.2, 154.4, 139.9, 133.8, 130.2, 129.4, 128.6, 128.5, 125.5, 120.1, 116.4, 80.3, 21.3; HRMS (ESI) calcd for $C_{16}H_{14}NaO_3$ $(M + Na)^+$ 277.0841, found 277.0839.

Compound 3es. 41 mg, 70% yield, colorless oil; ¹H NMR (600 MHz, CDCl3): δ 7.26−7.22 (m, 1H), 7.05−6.97 (m, 3H), 4.17 (d, J = 3.1 Hz, 1H), 4.08 (d, J = 3.1 Hz, 1H), 2.37 (s, 3H), 1.25 (s, 9H); ¹³C NMR (151 MHz, CDCl₃): δ 175.6, 156.2, 154.3, 139.8, 129.3, 125.4, 120.2, 116.5, 79.8, 39.0, 26.8, 21.3; HRMS (ESI) calcd for $C_{14}H_{18}NaO_3$ $(M + Na)^+$ 257.1154, found 257.1139.

Compound 3fa. 72 mg, 82% yield, colorless oil; ¹H NMR (600 MHz, CDCl₃): δ 8.13 (d, J = 7.3 Hz, 2H), 7.64 (t, J = 7.4 Hz, 1H), 7.52−7.49 (m, 2H), 7.27−7.24 (m, 1H), 7.10−7.07 (m, 2H), 6.94 (d, J $= 7.5$ Hz, 1H), 4.60 (d, J = 10.4 Hz, 1H), 2.39 (s, 3H), 2.22–2.12 (m, 1H), 1.52−1.45 (m, 2H), 1.39−1.28 (m, 6H), 1.00−0.95 (m, 6H); 13C NMR (151 MHz, CDCl₃): δ 163.6, 155.8, 149.1, 139.6, 133.7, 130.1, 129.2, 128.7, 128.5, 124.2, 118.5, 114.6, 105.0, 37.6, 35.0, 29.5, 28.5, 22.7, 21.4, 14.1, 11.9; HRMS (ESI) calcd for $C_{23}H_{28}NaO_3 (M + Na)^4$ 375.1936, found 375.1933.

Compound 3ga. 60 mg, 86% yield, colorless oil; ¹H NMR (600 MHz, CDCl₃): δ 8.13 (dd, J = 8.2, 1.1 Hz, 2H), 7.64 (t, J = 7.4 Hz, 1H), 7.51−7.48 (m, 2H), 7.42−7.37 (m, 2H), 7.28−7.26 (m, 2H), 7.15 (t, J = 7.4 Hz, 1H), 5.92−5.80 (m, 1H), 5.14−5.10 (m, 1H), 5.07−5.03 (m, 1H), 4.88 (t, J = 7.6 Hz, 1H), 2.85−2.81 (m, 2H); ¹³C NMR (151 MHz, CDCl₃): δ 163.4, 155.3, 149.6, 135.8, 133.8, 130.2, 129.6, 128.5, 128.4, 123.8, 118.2, 115.3, 97.2, 29.0; HRMS (ESI) calcd for $C_{18}H_{16}NaO_3$ $(M + Na)^+$ 303.0997, found 303.0993.

Compound 3ha. 61 mg, 82% yield, colorless oil; ¹H NMR (600 MHz, CDCl3): δ 8.12 (dd, J = 8.3, 1.2 Hz, 2H), 7.65−7.61 (m, 1H), 7.51−7.48 (m, 2H), 7.38−7.35 (m, 2H), 7.23 (dd, J = 8.7, 1.0 Hz, 2H), 7.12 (t, J = 7.4 Hz, 1H), 4.89 (t, J = 7.6 Hz, 1H), 2.09−2.05 (m, 2H), 1.44−1.35 (m, 4H), 0.92 (t, J = 7.2 Hz, 3H); 13C NMR (151 MHz, CDCl₃): δ 163.6, 155.6, 148.3, 133.8, 130.2, 129.5, 128.5, 128.5, 123.5, 117.7, 100.8, 31.4, 24.5, 22.1, 13.8; HRMS (ESI) calcd for $C_{19}H_{20}NaO_3$ $(M + Na)^+$ 319.1310, found 319.1309.

Compound 3ia. 70 mg, 85% yield, colorless oil; ¹H NMR (600 MHz, CDCl₃): δ 8.10 (d, J = 7.7 Hz, 2H), 7.64 (t, J = 7.4 Hz, 1H), 7.51−7.48 (m, 2H), 7.31 (d, J = 8.9 Hz, 2H), 7.17 (d, J = 8.9 Hz, 2H), 4.87 (t, J = 7.6 Hz, 1H), 2.08–2.04 (m, 2H), 1.42–1.34 (m, 4H), 0.91 (t, J = 7.2 Hz, 3H); ¹³C NMR (151 MHz, CDCl₃): δ 163.6, 154.3, 148.3, 134.0, 130.3, 129.6, 129.5, 128.7, 123.0, 119.2, 101.3, 31.4, 24.5, 22.2, 13.9; HRMS (ESI) calcd for $C_{19}H_{19}CNaO_3$ (M + Na)⁺ 353.0920, found 353.0904.

Compound 3ja. 69 mg, 77% yield, colorless oil; 1 H NMR (600 MHz, CDCl₃): δ 8.15–8.08 (m, 2H), 7.65 (t, J = 7.4 Hz, 1H), 7.52– 7.49 (m, 2H), 7.33−7.30 (m, 2H), 7.20−7.15 (m, 2H), 4.78 (d, J = 9.6 Hz, 1H), 2.22−2.16 (m, 1H), 1.81−1.61 (m, 5H), 1.30−1.15 (m, 5H); ¹³C NMR (151 MHz, CDCl₃): δ 163.7, 154.3, 147.2, 133.9, 130.2, 129.5, 128.6, 128.5, 128.4, 119.1, 106.9, 34.6, 33.1, 25.8, 25.6; HRMS (ESI) calcd for $C_{21}H_{21}CINaO_3 (M + Na)^+$ 379.1077, found 379.1073.

Compound 3ka. 64 mg, 95% yield, colorless oil; ¹H NMR (600 MHz, CDCl₃): δ 8.26–8.20 (m, 2H), 7.68 (t, J = 7.5 Hz, 1H), 7.56– 7.53 (m, 2H), 7.37 (d, J = 7.5 Hz, 2H), 7.27−7.24 (m, 2H), 7.14 (t, J = 7.4 Hz, 1H), 5.46 (s, 1H), 4.14 (q, J = 7.0 Hz, 2H), 1.46 (t, J = 7.0 Hz, 3H); ¹³C NMR (151 MHz, CDCl₃): δ 163.4, 152.8, 134.3, 134.0, 130.4, 128.8, 128.7, 128.5, 127.4, 125.8, 91.5, 65.5, 14.4; HRMS (ESI) calcd for $C_{17}H_{16}NaO_3$ $(M + Na)^+$ 291.0997, found 291.0987.

Compound 3ks. 54 mg, 87% yield, colorless oil; ¹H NMR (600 MHz, CDCl₃): δ 7.30−7.27 (m, 4H), 7.17−7.14 (m, 1H), 5.33 (s, 1H), 4.03 (q, J = 7.0 Hz, 2H), 1.41 (t, J = 7.0 Hz, 3H), 1.31 (s, 9H); 13 C NMR (151 MHz, CDCl₃): δ 175.1, 152.9, 134.3, 128.1, 127.7, 125.6, 91.0, 65.1, 38.9, 26.9, 14.3; HRMS (APCI) calcd for $C_{15}H_{21}O_3$ $(M + H)^+$ 249.1491, found 249.1488.

Compound 3la. 66 mg, 93% yield, colorless oil; ¹H NMR (600 MHz, CDCl₃): δ 8.19 (d, J = 7.8 Hz, 2H), 7.67 (t, J = 7.4 Hz, 1H), 7.55−7.52 (m, 2H), 7.32−7.29 (m, 2H), 6.95−6.91 (m, 2H), 5.41 (s, 1H), 4.11 (q, J = 7.0 Hz, 2H), 1.44 (t, J = 7.0 Hz, 3H); ¹³C NMR (151) MHz, CDCl₃): δ 163.2, 160.8 (d, J = 245.3 Hz), 152.4 (d, J = 1.8 Hz), 134.0, 130.2, 130.1 (d, $J = 3.3$ Hz), 128.7 (d, $J = 7.7$ Hz), 128.6, 128.5, 115.2 (d, J = 21.4 Hz), 90.4, 65.4, 14.3; HRMS (ESI) calcd for $C_{17}H_{15}FNaO_3$ $(M + Na)^+$ 309.0903, found 309.0898.

Compound 3ma. 67 mg, 90% yield, yellow oil; ¹H NMR (600 MHz, CDCl₃): δ 8.21 (d, J = 7.3 Hz, 2H), 7.67 (t, J = 7.5 Hz, 1H), 7.55−7.52 (m, 2H), 7.29 (d, J = 8.8 Hz, 2H), 6.80 (d, J = 8.8 Hz, 2H), 5.42 (s, 1H), 4.11 (q, J = 7.0 Hz, 2H), 3.76 (s, 3H), 1.44 (t, J = 7.0 Hz, 3H); ¹³C NMR (151 MHz, CDCl₃): δ 163.4, 157.6, 151.6, 133.8, 130.2, 128.7, 128.6, 128.4, 126.5, 113.8, 91.2, 65.3, 55.0, 14.3; HRMS (ESI) calcd for $C_{18}H_{18}NaO_4 (M + Na)^+$ 298.1205, found 298.1201.

Compound 3na. 34 mg, 71% yield, colorless oil; ¹H NMR (600 MHz, DMSO): δ 8.04–8.00 (m, 2H), 7.75 (t, J = 7.5 Hz, 1H), 7.60– 7.57 (m, 2H), 4.00 (d, $J = 3.6$ Hz, 1H), 3.97 (d, $J = 3.6$ Hz, 1H), 3.93 $(q, J = 7.0 \text{ Hz}, 2\text{H}), 1.27 \text{ (t, } J = 7.0 \text{ Hz}, 3\text{H})$; ¹³C NMR (151 MHz, DMSO): δ 163.7, 157.1, 134.8, 130.2, 129.6, 128.6, 73.1, 65.3, 14.4; HRMS (ESI) calcd for $C_{11}H_{12}NaO_3$ $(M + Na)^+$ 215.0684, found 215.0673.

General Procedure for the Ni-Catalyzed Coupling of 3 with **Boronic Acids.** To a mixture of 4a (37 mg, 0.3 mmol), $Ni(PCy_3)_2Cl_2$ (6.8 mg, 0.01 mmol), and K_3PO_4 (160 mg, 0.75 mmol) was added a solution of 3ks (62 mg, 0.25 mmol) in 1 mL of toluene under a nitrogen atmosphere. After stirring at 130 °C for 12 h, the reaction mixture was quenched with water, extracted with EtOAc, washed with brine, dried over anhydrous Na₂SO₄, and concentrated. Column chromatography on silica gel (petroleum ether/EtOAc = 50:1) gave 48
mg (yield: 86%) of 5a²¹ as a colorless oil. ¹H NMR (600 MHz, DMSO): δ 7.35–7.27 (m, 5H), 7.10 (t, J = 7.6 Hz, 2H), 7.02 (t, J = 7.3 Hz, 1H), 6.91 (d, J = [7.](#page-6-0)5 Hz, 2H), 5.92 (s, 1H), 3.99 (q, J = 6.9 Hz, 2H), 1.34 (t, J = 7.0 Hz, 3H); ¹³C NMR (151 MHz, DMSO): δ 156.4, 137.2, 136.7, 129.5, 129.1, 128.9, 128.7, 128.4, 125.6, 102.5, 63.8, 15.0.

Compound 5b. 43 mg, 71% yield, colorless oil; 1 H NMR (600 MHz, DMSO): δ 7.32−7.28 (m, 2H), 7.16−7.11 (m, 4H), 7.05−7.02 $(m, 1H)$, 6.91 (d, J = 7.4 Hz, 2H), 5.93 (s, 1H), 3.99 (q, J = 6.9 Hz, 2H), 1.34 (t, J = 6.9 Hz, 3H); ¹³C NMR (151 MHz, DMSO): δ 162.4 $(d, J = 245.9 \text{ Hz})$, 155.4, 137.1, 133.0 $(d, J = 3.3 \text{ Hz})$, 131.7 $(d, J = 8.4 \text{ Hz})$ Hz), 129.0, 128.5, 125.7, 115.7 (d, J = 21.6 Hz), 102.6, 63.9, 15.0; HRMS (APCI) calcd for $C_{16}H_{16}FO (M + H)^+$ 243.1185, found 243.1180.

Compound 5c. 44 mg, 74% yield, colorless oil; 1 H NMR (600 MHz, DMSO): δ 7.16 (d, J = 8.1 Hz, 2H), 7.13–7.08 (m, 4H), 7.01 (t, $J = 7.3$ Hz, 1H), 6.92 (d, $J = 7.3$ Hz, 2H), 5.87 (s, 1H), 3.96 (q, $J = 6.9$ Hz, 2H), 2.30 (s, 3H), 1.33 (t, J = 6.9 Hz, 3H); ¹³C NMR (151 MHz, DMSO): δ 156.4, 138.4, 137.4, 133.7, 129.4, 129.3, 128.9, 128.4, 125.5, 102.2, 63.7, 21.4, 15.0; HRMS (APCI) calcd for $C_{17}H_{19}O (M + H)^+$ 239.1436, found 239.1421.

Compound 5d. 42 mg, 70% yield, colorless oil; ¹H NMR (600 MHz, DMSO): δ 7.32–7.26 (m, 2H), 7.18–7.16 (m, 1H), 7.11 (d, J = 7.5 Hz, 1H), 7.05−7.02 (m, 2H), 6.99−6.94 (m, 1H), 6.77−6.72 (m, 2H), 5.97 (s, 1H), 3.99 (q, J = 7.0 Hz, 2H), 2.16 (s, 3H), 1.31 (t, J = 7.0 Hz, 3H); 13C NMR (151 MHz, DMSO): δ 156.6, 136.9, 136.7, 136.6, 130.7, 129.9, 129.2, 128.4, 127.9, 126.5, 125.4, 103.2, 63.5, 19.3, 15.0; HRMS (APCI) calcd for $C_{17}H_{19}O(M + H)^+$ 239.1436, found 239.1425.

Compound 5e. 61 mg, 83% yield, colorless oil; 1 H NMR (600 MHz, DMSO): δ 7.64 (d, J = 8.3 Hz, 2H), 7.48 (d, J = 8.2 Hz, 2H), 7.14−7.11 (m, 2H), 7.06 (t, J = 7.3 Hz, 1H), 6.94 (d, J = 7.6 Hz, 2H), 6.04 (s, 1H), 4.01 (q, J = 6.9 Hz, 2H), 1.34 (t, J = 6.9 Hz, 3H); ¹³C NMR (151 MHz, DMSO): δ 154.7, 140.7 (q, J = 1.2 Hz), 136.7, 130.3, 129.3 (q, J = 31.5 Hz), 129.2, 128.6, 126.0, 125.5 (q, J = 3.8 Hz), 124.5 (q, J = 272.4 Hz), 103.8, 64.1, 14.9; HRMS (APCI) calcd for $C_{17}H_{16}F_3O (M + H)^+$ 293.1153, found 293.1151.

Compound 5f. 40 mg, 60% yield, colorless oil; 1 H NMR (600 MHz, DMSO): δ 7.45 (d, J = 7.3 Hz, 2H), 7.39−7.36 (m, 2H), 7.32− 7.29 (m, 1H), 7.11 (t, J = 7.8 Hz, 1H), 6.80 (s, 1H), 6.76−6.73 (m, 2H), 5.36 (t, J = 7.7 Hz, 1H), 2.24−2.19 (m, 5H), 1.40−1.26 (m, 4H), 0.83 (t, J = 7.3 Hz, 3H); ¹³C NMR (151 MHz, DMSO): δ 157.4, 150.1, 139.6, 134.6, 129.7, 128.8, 128.8, 128.7, 123.2, 118.2, 117.0, 114.6, 32.2, 26.8, 22.2, 21.4, 14.2; HRMS (APCI) calcd for $C_{19}H_{23}O$ $(M + H)^+$ 267.1749, found 267.1745.

Compound 5g. 40 mg, 63% yield, colorless oil; ¹H NMR (600 MHz, DMSO): δ 7.53−7.49 (m, 2H), 7.34−7.31 (m, 2H), 7.29−7.26 $(m, 1H)$, 7.13 (t, J = 7.8 Hz, 1H), 6.81–6.76 $(m, 2H)$, 6.71 (dd, J = 8.2, 2.3 Hz, 1H), 6.07 (t, J = 7.4 Hz, 1H), 5.88−5.81 (m, 1H), 5.11− 5.06 (m, 1H), 5.03−5.00 (m, 1H), 2.91−2.89 (m, 2H), 2.23 (s, 3H); 13C NMR (151 MHz, DMSO): ^δ 157.3, 148.9, 139.8, 136.4, 135.2, 129.9, 129.1, 128.6, 125.6, 122.9, 116.3, 116.0, 115.4, 112.6, 30.2, 21.5; HRMS (APCI) calcd for $C_{18}H_{19}O$ $(M + H)^+$ 251.1436, found 251.1434.

Compound 5h. 26 mg, 50% yield, colorless oil; ¹H NMR (600 MHz, DMSO): δ 7.70−7.66 (m, 2H), 7.44−7.37 (m, 3H), 7.26 (t, J = 7.8 Hz, 1H), 6.96−6.86 (m, 3H), 5.28 (d, J = 2.4 Hz, 1H), 4.48 (d, J = 2.3 Hz, 1H), 2.30 (s, 3H); 13C NMR (151 MHz, DMSO): δ 158.6, 156.3, 140.0, 135.0, 130.0, 129.5, 129.0, 125.8, 124.7, 120.0, 116.5, 94.3, 21.4; HRMS (APCI) calcd for $C_{15}H_{15}O(M + H)^+$ 211.1123, found 211.1105.

■ ASSOCIATED CONTENT

S Supporting Information

Spectroscopic data of products 3 and 5. This material is available free of charge via the Internet at http://pubs.acs.org.

■ AUTHOR INFORMATION

Corresponding Author

*E-mail: gangguo@zjnu.cn (G.Z.).

Notes

The auth[ors declare no com](mailto:gangguo@zjnu.cn)peting financial interest.

■ ACKNOWLEDGMENTS

We are grateful for the financial support from the Natural Science Foundation of Zhejiang Province (LR12B02001 and LY14B020002), the National Science Foundation of China (21172199), and Zhejiang Normal University.

■ REFERENCES

(1) For selected reviews, see: (a) Bruneau, C.; Dixneuf, P. H. Chem. Commun. 1997, 507. (b) Dixneuf, P. H.; Bruneau, C.; Dérien, S. Pure Appl. Chem. 1998, 70, 1065. (c) Bruneau, C.; Dixneuf, P. H. Acc. Chem. Res. 1999, 32, 311. (d) Bruneau, C.; Dixneuf, P. H. Angew. Chem., Int. Ed. 2006, 45, 2176. (e) Alonso, F.; Beletskaya, I. P.; Yus, M. Chem. Rev. 2004, 104, 3079. (f) Gooßen, L. J.; Rodríguez, N.; Gooßen, K. Angew. Chem., Int. Ed. 2008, 47, 3100.

(2) For intramolecular versions, see: (a) Chan, D. M. T.; Marder, T. B.; Milstein, D.; Taylor, N. J. J. Am. Chem. Soc. 1987, 109, 6385. (b) Elgafi, S.; Field, L. D.; Messerle, B. A. J. Organomet. Chem. 2000, 607, 97. (c) Mas-Marzá, E.; Sanaú, M.; Peris, E. Inorg. Chem. 2005, 44, 9961. (d) Li, X.; Chianese, A. R.; Vogel, T.; Crabtree, R. H. Org. Lett. 2005, 7, 5437. (e) Harkat, H.; Weibel, J.-M.; Pale, P. Tetrahedron Lett. 2006, 47, 6273. (f) Genin, E.; Toullec, P. Y.; Antoniotti, S.; Brancour, C.; Genêt, J.-P.; Michelet, V. J. Am. Chem. Soc. 2006, 128, 3112. For a review, see: (g) Drozdzak, R.; Allaert, B.; Ledoux, N.; Dragutan, V.; Verpoort, F. Adv. Synth. Catal. 2005, 347, 1721.

(3) (a) Lemaire, H.; Lucas, H. J. J. Am. Chem. Soc. 1955, 77, 939. (b) Hudrlik, P. F.; Hudrlik, A. M. J. Org. Chem. 1973, 38, 4254. (c) Larock, R. C.; Oertle, K.; Beatty, K. M. J. Am. Chem. Soc. 1980, 102, 1966. (d) Kita, Y.; Akai, S.; Ajimura, N.; Yoshigi, M.; Tsugoshi, T.; Yasuda, H.; Tamura, Y. J. Org. Chem. 1986, 51, 4150.

(4) (a) Shvo, Y.; Rotem, M. Organometallics 1983, 2, 1689. (b) Mitsudo, T.-a.; Hori, Y.; Yamakawa, Y.; Watanabe, Y. J. Org. Chem. 1987, 52, 2230. (c) Doucet, H.; Martin-Vaca, B.; Bruneau, C.; Dixneuf, P. H. J. Org. Chem. 1995, 60, 7247. (d) Goossen, L. J.; Paetzold, J.; Koley, D. Chem. Commun. 2003, 706. (e) Kim, H.; Goble, S. D.; Lee, C. J. Am. Chem. Soc. 2007, 129, 1030. (f) Yi, C. S.; Gao, R. Organometallics 2009, 28, 6585. (g) Tanaka, K.; Saitoh, S.; Hara, H.; Shibata, Y. Org. Biomol. Chem. 2009, 7, 4817. (h) Tripathy, J.; Bhattacharjee, M. Tetrahedron Lett. 2009, 50, 4863. (i) Willem, Q.; Nicks, F.; Sauvage, X.; Delaude, L.; Demonceau, A. J. Organomet. Chem. 2009, 694, 4049. (j) Nicks, F.; Aznar, R.; Sainz, D.; Muller, G.; Demonceau, A. Eur. J. Org. Chem. 2009, 5020. (k) Berger, S.; Haak, E. Tetrahedron Lett. 2010, 51, 6630. (l) Nishiumi, M.; Miura, H.; Wada, K.; Hosokawa, S.; Inoue, M. Adv. Synth. Catal. 2010, 352, 3045. (m) Tan, S. T.; Fan, W. Y. Eur. J. Inorg. Chem. 2010, 4631. (n) Cadierno, V.; Francos, J.; Gimeno, J. Organometallics 2011, 30, 852.

(5) Lumbroso, A.; Vautravers, N. R.; Breit, B. Org. Lett. 2010, 12, 5498.

(6) Nakagawa, H.; Okimoto, Y.; Sakaguchi, S.; Ishii, Y. Tetrahedron Lett. 2003, 44, 103.

(7) (a) Lambert, C.; Utimoto, K.; Nozaki, H. Tetrahedron Lett. 1984, 25, 5323. (b) Wakabayashi, T.; Ishii, Y.; Murata, T.; Mizobe, Y.; Hidai, M. Tetrahedron Lett. 1995, 36, 5585. (c) Wolf, L. B.; Tjen, K. C. M. F.; ten Brink, H. T.; Blaauw, R. H.; Hiemstra, H.; Schoemaker, H. E.; Rutjes, F. P. J. T. Adv. Synth. Catal. 2002, 344, 70. (d) Wang, Y.; Burton, D. J. J. Org. Chem. 2006, 71, 3859. (e) Lu, X.; Zhu, G.; Ma, S. Tetrahedron Lett. 1992, 33, 7205. (f) Zhang, Q.; Xu, W.; Lu, X. J. Org. Chem. 2005, 70, 1505. (g) Zhao, L.; Lu, X.; Xu, W. J. Org. Chem. 2005, 70, 4059. (h) Song, J.; Shen, Q.; Xu, F.; Lu, X. Tetrahedron 2007, 63, 5148. (i) Tang, D.-J.; Tang, B.-X.; Li, J.-H. J. Org. Chem. 2009, 74, 6749.

(8) Ishino, Y.; Nishiguchi, I.; Nakao, S.; Hirashima, T. Chem. Lett. 1981, 641.

(9) (a) Chary, B. C.; Kim, S. J. Org. Chem. 2010, 75, 7928. (b) Wang, Y.; Wang, Z.; Li, Y.; Wu, G.; Cao, Z.; Zhang, L. Nat. Commun. 2014, DOI: 10.1038/ncomms4470.

(10) Hua, R.; Tian, X. J. Org. Chem. 2004, 69, 5782.

(11) For a DABCO-promoted addition of carboxylates to electrondeficient alkynes, see: Fan, M.-J.; Li, G.-Q.; Liang, Y.-M. Tetrahedron 2006, 62, 6782.

(12) Kawatsura, M.; Namioka, J.; Kajita, K.; Yamamoto, M.; Tsuji, H.; Itoh, T. Org. Lett. 2011, 13, 3285.

(13) Smith, D. L.; Goundry, W. R. F.; Lam, H. W. Chem. Commun. 2012, 48, 1505.

The Journal of Organic Chemistry Article and the Second Secon

(14) Tsukada, N.; Takahashi, A.; Inoue, Y. Tetrahedron Lett. 2011, 52, 248.

(15) Cai, H.; Yuan, Z.; Zhu, W.; Zhu, G. Chem. Commun. 2011, 47, 8682.

(16) (a) Bai, Y.; Yin, J.; Kong, W.; Mao, M.; Zhu, G. Chem. Commun. 2013, 49, 7650. (b) Cui, W.; Ying, J.; Zheng, R.; Cheng, C.; Bai, Y.; Zhu, G. J. Org. Chem. 2014, 79, 3487.

(17) Cui, W.; Mao, M.; He, Z.; Zhu, G. J. Org. Chem. 2013, 78, 9815. (18) For the synthesis of ynol ethers employed in this paper, see: (a) Jouvin, K.; Bayle, A.; Legrand, F.; Evano, G. Org. Lett. 2012, 14, 1652. (b) Tanaka, R.; Miller, S. I. Tetrahedron Lett. 1971, 12, 1753. (c) Moyano, A.; Charbonnier, F.; Greene, A. E. J. Org. Chem. 1987, 52, 2919. For recent reports on ynol ethers, see: (d) Miyauchi, Y.; Noguchi, K.; Tanaka, K. Org. Lett. 2012, 14, 5856. (e) Tran, V.; Minehan, T. G. Org. Lett. 2012, 14, 6100. (f) Verrier, C.; Carret, S.; Poisson, J.-F. Org. Lett. 2012, 14, 5122. (g) Minami, Y.; Shiraishi, Y.; Yamada, K.; Hiyama, T. J. Am. Chem. Soc. 2012, 134, 6124. (h) Minami, Y.; Yamada, K.; Hiyama, T. Angew. Chem., Int. Ed. 2013, 52, 10611. (i) Graf, K.; Rü hl, C. L.; Rudolph, M.; Rominger, F.; Hashmi, A. S. K. Angew. Chem., Int. Ed. 2013, 52, 12727. (j) Alford, J. S.; Davies, H. M. L. J. Am. Chem. Soc. 2014, 136, 10266. For a recent review, see: (k) Evano, G.; Gaumont, A.-C.; Alayrac, C.; Wrona, I. E.; Giguere, J. R.; Delacroix, O.; Bayle, A.; Jouvin, K.; Theunissen, C.; Gatignol, J.; Silvanus, A. C. Tetrahedron 2014, 70, 1529 and references therein.

(19) Miyaura, N.; Suzuki, A. Chem. Rev. 1995, 95, 2457.

(20) For the Ni-catalyzed Suzuki−Miyaura coupling of C−OPiv bonds, see: (a) Guan, B.-T.; Wang, Y.; Li, B.-J.; Yu, D.-G.; Shi, Z.-J. J. Am. Chem. Soc. 2008, 130, 14468. (b) Sun, C.-L.; Wang, Y.; Zhou, X.; Wu, Z.-H.; Li, B.-J.; Guan, B.-T.; Shi, Z.-J. Chem.-Eur. J. 2010, 16, 5844. (c) Quasdorf, K. W.; Tian, X.; Garg, N. K. J. Am. Chem. Soc. 2008, 130, 14422. (d) Ramgren, S. D.; Hie, L.; Ye, Y.; Garg, N. K. Org. Lett. 2013, 15, 3950. (e) Zhou, Q.; Srinivas, H. D.; Dasgupta, S.; Watson, M. P. J. Am. Chem. Soc. 2013, 135, 3307. For recent reviews on the cross-coupling of C−O bonds, see: (f) Li, B.-J.; Yu, D.-G.; Sun, C.-L.; Shi, Z.-J. Chem.-Eur. J. 2011, 17, 1728. (g) Rosen, B. M.; Quasdorf, K. W.; Wilson, D. A.; Zhang, N.; Resmerita, A.-M.; Garg, N. K.; Percec, V. Chem. Rev. 2011, 111, 1346. (h) Yamaguchi, J.; Muto, K.; Itami, K. Eur. J. Org. Chem. 2013, 19.

(21) Caputo, R.; Longobardo, L.; Palumbo, G.; Pedatella, S. Synlett 1995, 1274. Of note, the hydrolysis of 5 was observed in CDCl₃; as such, DMSO-d6 was used for the NMR measurements.

(22) So far, the one-pot hydrocarboxylation/Suzuki−Miyaura coupling process was unsuccessful.

(23) Burke, M. D.; Schreiber, S. L. Angew. Chem., Int. Ed. 2004, 43, 46.